Two-orbital degeneracy lifted local precursor to a metal-insulator transition in MgTi₂O₄

Long Yang[®],¹ Robert J. Koch[®],² Hong Zheng,³ J. F. Mitchell[®],³ Weiguo Yin[®],² Matthew G. Tucker[®],⁴ Simon J. L. Billinge[®],^{1,2} and Emil S. Bozin[®],^{*}

¹Department of Applied Physics and Applied Mathematics, Columbia University, New York, New York 10027, USA

²Condensed Matter Physics and Materials Science Division, Brookhaven National Laboratory, Upton, New York 11973, USA

³Materials Science Division, Argonne National Laboratory, Argonne, Illinois 60439, USA

⁴Neutron Scattering Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37830, USA

(Received 17 September 2020; revised 6 November 2020; accepted 6 November 2020; published 14 December 2020)

The MgTi₂O₄ spinel exhibits a metal-insulator transition on cooling below $T_s \approx 250$ K, accompanied by Ti t_{2g}^{1} orbital ordering and spin-singlet dimerization with associated average symmetry reduction to tetragonal. By combining x-ray and neutron pair distribution function analyses to track the evolution of the local atomic structure across the transition we find that *local* tetragonality already exists in the metallic *globally* cubic phase at high temperature. Local distortions are observed up to at least 500 K, the highest temperature assessed in this study. Significantly, the high-temperature local state is not continuously connected to the orbitally ordered band insulator ground state and so the transition cannot be characterized as a trivial order-disorder type. The shortest Ti-Ti spin-singlet dimer bond lengths expand abruptly on warming across the transition, but remain shorter than those seen in the cubic average structure. These seemingly contradictory observations can be understood within the model of a local fluctuating two-orbital t_{2g} orbital degeneracy lifted (ODL) precursor state derived from electron filling, Ti substructure topology, and point symmetry considerations. The ODL state in $MgTi_2O_4$ has a correlation length of about 1 nm at high temperature. We discuss that this extended character of the local distortions is consistent with the two-orbital nature of this state imposed by the charge filling and the bond charge repulsion. The MgTi₂O₄ spinel exemplifies multiorbital ODL state and presents the possibility of a widespread presence of such precursor states in scarcely studied high-temperature regimes of transition-metal-based quantum materials.

DOI: 10.1103/PhysRevB.102.235128

I. INTRODUCTION

The AB_2X_4 transition-metal spinels, with B cations forming a geometrically frustrated pyrochlore lattice [1], exhibit a wealth of interesting electronic and magnetic properties, such as spin dimerization [2,3], spin-lattice instability [4], orbital ordering [5], charge ordering [6], and metal-insulator transitions [7-9]. For instance, the CuIr₂S₄ thiospinel system exhibits a transition from paramagnetic metal to diamagnetic insulator on cooling [10-12], the formation of structural isomorphic octamers [2,13], and anomalous electrical properties [14,15]. In particular, CuIr₂S₄ has broken symmetries in its ground state accompanied by the formation of charge order and orbital order and the creation of magnetic spin singlets on dimerized Ir^{4+} - Ir^{4+} pairs [2]. These dimers were shown to disappear on warming through the transition, but preformed local symmetry broken states were seen at temperatures well above the global long-range-ordering structure transition [16]. Thus, on cooling, a precursor state exists that has broken local symmetry but is high symmetry over the long range. This behavior was shown to be driven by breaking of *d*-electron orbital degeneracies, resulting in a local fluctuating orbital degeneracy lifted (ODL) state distinct from the spin-singlet dimer. This local state results from direct t_{2g} Ir orbital overlap promoted by the topology of the crystal structure. In the regime of partial filling and high crystal symmetry that imposes the degeneracy of the orbital manifold, a molecular-orbital-like state is formed, accompanied by local structure distortion. Many of the interesting physical properties of CuIr₂S₄ could be explained by the short-range-ordered ODL mechanism, such as the nonconventional conduction in the high-temperature metallic phase [14] and the apparently contradictory observation of the destruction of the dimers at the phase transition but the persistence of poor metallic response above the transition [17].

Similar local symmetry-broken ODL states preformed at high temperature have been observed in other nonspinel *d*electron systems such as the FeSe superconductor [18,19]. Even the well-studied physics of the perovskite lanthanum manganites (LaMnO₃) could be interpreted in the same way [20], arguing that the ODL phenomenon may be widespread among the myriad of materials with partially filled *d*-electron manifolds. To explore this hypothesis it is important to seek it out systematically and characterize other materials that have the potential of revealing both commonalities and novel aspects that give more insights into the general ODL phenomenon. For instance, there is no fundamental reason for the ODL states to be exclusively comprised of one orbital per transition metal, yet multiorbital ODL states have so far not

^{*}bozin@bnl.gov

been identified. Here we show that such a multiorbital ODL state exists in a related spinel, $MgTi_2O_4$, and that this is a consequence of the d^1 -electron configuration decorating the pyrochlore lattice.

MgTi₂O₄ shares many features with CuIr₂S₄ [21]. They are both cubic spinels at high temperature with a transitionmetal ion pyrochlore sublattice of corner-shared tetrahedra. In both materials the crystal field splits the d orbitals into a t_{2g} triplet and an e_g doublet with the t_{2g} orbitals partially occupied and the e_g orbitals empty. They both exhibit a temperaturedependent metal-insulator transition (MIT) [8,22], the origin of which has been attributed to an orbital-selective Peierls mechanism [21]. They both exhibit a global symmetry lowering, from cubic to tetragonal, at the MIT on cooling. They both have anomalous electrical resistivity behavior in the high-temperature metallic phase [14,23]. The symmetry lowering at the MIT is accompanied by a dimerization of transition-metal ions that results in alternating short and long metal-metal bonds and a resulting tetramerization [24], along linear chains of ions on the pyrochlore sublattice [21]. The short bonds are associated with spin-singlet dimer formation [2,3]. The charge filling is also electron-hole symmetric between the systems with Ti^{3+} having one electron in the t_{2g} manifold, while Ir⁴⁺ has one hole, although the nominal charge of Ir in $CuIr_2S_4$ is 3.5+, which would place half a hole per Ir in the t_{2g} antibonding band on average in this compound.

Despite the similarities, there are also notable differences. The Ti valence electrons reside in 3d orbitals whereas for Ir they are 5d, which are more extended and should result in larger bandwidth. Indeed, the average separation between the transition metals on the undistorted tetrahedral pyrochlore sublattices of their cubic structures follows this expectation: Ti-Ti separation is shorter (order of 3.0 Å) than Ir-Ir separation (order of 3.5 Å). Also, experimentally, in MgTi₂O₄ the tetragonal distortion is shown to be compressive (c < a) below the MIT [3] while it is tensile (c > a) [10] in CuIr₂S₄. Dimers form helical superstructures in MgTi₂O₄, whereas they form octameric "molecules" in CuIr₂S₄, thereby lowering the symmetry further to triclinic. Both materials tetramerize below the MIT; however, $CuIr_2S_4$ has a 3+-3+-4+-4+ charge ordering (CO) that accompanies an orbital ordering [16], whereas a uniform 3+ charge on Ti rules out CO in MgTi₂O₄. With both similarities (t_{2g} orbitals on pyrochlore lattice and formation of dimerized singlets) to and differences from CuIr₂S₄ (lack of charge order and electron states rather than hole states), MgTi₂O₄ provides a natural next step in a deeper, broader mapping of ODL phenomena. As we show below, this includes the emergence of a multiorbital ODL state.

Neutron pair distribution function (PDF) analysis on $MgTi_2O_4$ has been performed previously [25] to study the spin-singlet dimers, suggesting that they do persist to high temperature. However, due to the weak and negative neutron scattering length of Ti and appreciable overlap with substantially stronger oxygen signal, neutron PDF analysis itself cannot fully reveal how the local structure behaves with temperature. Here we have applied a combined x-ray and neutron analysis to understand the full picture. We find unambiguously that the Ti-Ti dimers do disassemble on warming through the MIT. However, the local structure does not agree with the average structure even at high temperature. In analogy

with CuIr₂S₄, partially filled t_{2g} transition-metal orbital manifolds of MgTi₂O₄, which are triply degenerate in the average cubic symmetry, utilize their favorable overlaps fostered by the pyrochlore sublattice topology to form an ODL state. Its structural signatures are observed up to at least 500 K (order of $2T_s$), the highest temperature measured. The spatial extent of the local structural response is found to be greater than that observed in CuIr₂S₄, consistent with the proposed two-orbital character of the ODL state in MgTi₂O₄.

II. METHODS

A. Sample preparation and characterization

TiO₂, Ti metal, and an excess of MgO were mixed and reacted using a spark plasma sintering technique in a graphite crucible. Synthesis at 1100 °C was complete in approximately 15 min. The sample was reground and fired a second time under similar conditions. Powder x-ray diffraction analysis of the product showed well-crystallized MgTi₂O₄ spinel accompanied by an extremely small concentration of Ti₂O₃ as a second phase. Magnetization measurements were conducted using a superconducting quantum interference device magnetometer on a specimen with mass of 6.2 mg. The data show a pronounced low-temperature Curie tail which was subtracted. The Curie-Weiss fit to the low-temperature data yielded a Weiss temperature of -0.45 K, consistent with isolated spins, and a Curie constant of 0.023 emu K/mol, which corresponds to approximately 3 mol % of putative Ti³⁺ spin-1/2 impurities.

B. The PDF method

The local structure was studied using the atomic PDF technique [26,27]. The PDF analysis of x-ray and neutron powder diffraction data sets has been demonstrated to be an excellent tool for revealing local structural distortions in many systems [16,18,20,28–31]. The PDF gives the scaled probability of finding two atoms in a material a distance r apart and is related to the density of atom pairs in the material. It does not presume periodicity and so goes well beyond just well-ordered crystals [26,27]. The experimental PDF, denoted by G(r), is the truncated Fourier transform of the reduced total scattering structure function [32] F(Q) = Q[S(Q) - 1],

$$G(r) = \frac{2}{\pi} \int_{Q_{\min}}^{Q_{\max}} F(Q) \sin(Qr) dQ, \qquad (1)$$

where Q is the magnitude of the scattering momentum transfer. The total scattering structure function S(Q) is extracted from the Bragg and diffuse components of x-ray, neutron, or electron powder diffraction intensity. For elastic scattering, $Q = 4\pi \sin(\theta)/\lambda$, where λ is the wavelength of the probe and 2θ is the scattering angle. In practice, values of Q_{\min} and Q_{\max} are determined by the experimental setup and Q_{\max} is often reduced below the experimental maximum to eliminate noisy data from the PDF since the signal-to-noise ratio becomes unfavorable in the high-Q region [26].

C. X-ray PDF experiment

The synchrotron x-ray total scattering measurements were carried out at the PDF beamline (28-ID-1) at the National Synchrotron Light Source II (NSLS-II) at Brookhaven National Laboratory (BNL) using the rapid acquisition PDF method (RAPDF) [33]. The MgTi₂O₄ powder sample was loaded in a 1-mm-diam polymide capillary and measured from 90 to 500 K on warming using a flowing nitrogen cryostream provided by Oxford Cryosystems 700 Series Cryocooler. The experimental setup was calibrated by measuring the crystalline Ni as a standard material. A two-dimensional PerkinElmer area detector was mounted behind the samples perpendicular to the primary beam path with a sampleto-detector distance of 227.7466 mm. The incident x-ray wavelength was 0.1668 Å. The PDF instrument resolution effects are accounted for by two parameters in modeling, Q_{damp} and Q_{broad} [34,35]. For x-ray PDF measurement, these were determined as $Q_{\text{damp}} = 0.039 \text{ Å}^{-1}$ and $Q_{\text{broad}} = 0.010 \text{ Å}^{-1}$ by fitting the x-ray PDF from a well-crystallized sample of Ni collected under the same experimental conditions.

In order to verify the data reproducibility, an additional set of x-ray data (in the 90–300 K temperature range on warming) was collected at the XPD beamline (28-ID-2) at the NSLS-II at BNL using a similar RAPDF setup but with x-ray wavelength of 0.1901 Å and sample-to-detector distance of 251.1493 mm. The corresponding instrument resolution parameters were determined to be $Q_{damp} = 0.032$ Å⁻¹ and $Q_{broad} = 0.010$ Å⁻¹, implying similar instrument resolution effects across the two sets of measurements.

The collected x-ray data frames were summed, corrected for detector and polarization effects, and masked to remove outlier pixels before being integrated along arcs of constant momentum transfer Q, to produce one-dimensional powder diffraction patterns using the PYFAI program [36]. Standardized corrections and normalizations were applied to the data to obtain the reduced total scattering structure function F(Q), which was Fourier transformed to obtain the PDF, using PDFGETX3 [37] within xPDFSUITE [38]. The maximum range of data used in the Fourier transform was chosen to be $Q_{max} =$ 25.0 Å⁻¹ so as to give the best trade-off between statistical noise and real-space resolution.

D. Neutron PDF experiment

The time-of-flight (TOF) neutron total scattering measurements were conducted at the NOMAD beamline (BL-1B) [39] at the Spallation Neutron Source at Oak Ridge National Laboratory. The MgTi₂O₄ powder sample was loaded into a 3-mm-diam silica capillary mounted on a sample shifter and data were collected from 100 to 500 K on warming using an Ar flow cryostream. The neutron PDF instrument resolution parameters were determined as $Q_{damp} = 0.0232$ Å⁻¹ and $Q_{broad} = 0.0175$ Å⁻¹ by fitting the neutron PDF of a NIST standard 640d crystalline silicon sample. The neutron data were reduced and transformed to the PDF with $Q_{max} = 25.0$ Å⁻¹ using the automated data reduction scripts available at the NOMAD beamline.

E. Structural modeling

The PDF modeling programs PDFGUI and DIFFPY-CMI were used for PDF structure refinements [35,40]. In these refinements U_{iso} (Å²) is the isotropic atomic displacement

parameter (ADP) and the ADPs of the same type of atoms are constrained to be the same; δ_2 (Å²) is a parameter that describes correlated atomic motions [41]. The PDF instrument parameters Q_{damp} and Q_{broad} determined by fitting the PDF from the well-crystallized standard sample under the same experimental conditions are fixed in the structural refinements on the MgTi₂O₄ data set.

To estimate the length scale of local structural correlations, an r_{\min} -dependent fit is performed on select high-temperature x-ray data (300, 400, and 500 K). During the fit, the average high-temperature cubic MgTi₂O₄ model is used, fixing the upper limit of fit range as $r_{\max} = 50$ Å, but changing the lower limit r_{\min} from $1 \le r_{\min} \le 36$ Å in 0.2-Å r steps. Each fit range uses the same initial parameter values and the δ_2 term is not applied since the data fitting range does not consistently include the low-r region in all these refinements.

The Rietveld refinement on the neutron TOF Bragg data was implemented in the GSAS-II software [42]. The instrument parameters were refined to the standard silicon data collected under the same experimental conditions and then fixed in the MgTi₂O₄ Rietveld refinements. The sequential refinement option was used to refine the temperature series data set collected at the $2\theta = 120^{\circ}$ detector bank from 100 to 500 K in a systematic manner.

F. Structural models

Two candidate MgTi₂O₄ models were fit against the experimental data. In the cubic MgTi₂O₄ model (space group Fd3m), the atoms sit at the following Wyckoff positions: Mg at 8a (0.125,0.125,0.125), Ti at 16d (0.5, 0.5, 0.5), and O at 32e (x, x, x). The initial lattice parameters and atomic positions are a = 8.509027 Å and O at (0.25920, 0.25920, 0.25920) [3]. The Ti pyrochlore sublattice is shown in Fig. 1(a), indicating that all the Ti-Ti bonds are of equal length, reflecting regular Ti₄ tetrahedra. In the tetragonal MgTi₂O₄ model (space group $P4_12_12$), the atoms sit at the following Wyckoff positions: Mg at 4a (x, x, 0), Ti at 8b (x, y, z), O1 at 8b (x, y, z), and O2 at 8b (x, y, z). The initial lattice parameters and atomic positions are a = 6.02201 Å, c = 8.48482 Å, Mg at (0.7448, 0.7448, 0), Ti at (-0.0089, 0.2499, -0.1332), 01 at (0.4824, 0.2468, 0.1212),and O2 at (0.2405, 0.0257, 0.8824) [3]. The corresponding distorted Ti sublattice is presented in Fig. 1(b), showing that one Ti-Ti bond gets shorter (indicated in red) and one gets longer (in yellow) out of the six Ti-Ti bonds of each Ti₄ tetrahedron. The lattice parameters of these two models have the relationship $c_c \sim c_t \sim \sqrt{2a_t}$, where the subscripts c and t refer to the cubic and tetragonal models, respectively.

III. VANISHING SPIN-SINGLET DIMERS

A. Canonical behavior

The magnetization M(T) of the sample up to 300 K is shown in Fig. 1(c). A low-temperature Curie-Weiss-like component has been subtracted to account for the effect of magnetic impurities contributing to the signal at low temperature. The onset of a broad dimerization transition is observed at around 250 K, which is close to the literature-reported



FIG. 1. The Ti pyrochlore sublattice of corner-shared Ti₄ tetrahedra for (a) undistorted (cubic) and (b) distorted (tetragonal) MgTi₂O₄ structures, highlighting both short (red) and long (yellow) Ti-Ti bonds in the distorted structure. (c) Temperature dependence of magnetization with the Curie-Weiss behavior contribution subtracted. The inset shows a TiO₆ octahedron and the t_{2g} orbitals that point towards O-O edges. The Ti atom is in blue and the O atom is in red. The vertical gray dashed line at 250 K marks the MIT temperature.

MIT temperature of 260 K [8,43] and implies the canonical behavior of our sample.

We next establish that the average structural behavior of our sample agrees with other observations in the literature [3,8,44] by carrying out Rietveld refinements fitting tetragonal and cubic MgTi₂O₄ models to the neutron TOF Bragg data measured from 100 to 500 K. The refinement results from select fits are reproduced in Table I presented in the Appendix, and a representative fit, of the tetragonal model to the 100 K data set, is shown in Fig. 2(a). The tetragonal distortion, given by $c/\sqrt{2}a$, is small at all temperatures and the data are not of high enough resolution to directly observe distinct peaks that would indicate any tetragonal distortion. However, the temperature dependence of the weighted profile agreement factor R_{wp} of the two models is shown in Fig. 2(b), which clearly implicates the tetragonal model as preferred at low temperature, but not at high temperature, above around 230–250 K, which is close to the literature-reported MIT temperature 260 K [8].

B. Local structure behavior

We investigate the behavior of the local structure by performing a PDF structural refinement of the x-ray total scattering PDF data from 90 to 500 K. Fits were carried out over the data ranges of 1.5 < r < 4 Å and 1.5 < r < 10 Å and the results are qualitatively the same. A representative fit of the



FIG. 2. (a) The 100 K neutron powder diffraction pattern (blue) collected using the $2\theta = 120^{\circ}$ detector bank fit by the tetragonal MgTi₂O₄ model (red) with the difference curve (green) offset below. (b) Resulting neutron Rietveld refinement weighted profile agreement factor R_{wp} values versus temperature on neutron TOF Bragg data using tetragonal (blue) and cubic (red) MgTi₂O₄ models from 100 to 500 K (120° detector bank data). (c) The x-ray PDF of 100 K data (blue) fit by the tetragonal model (red) over the range 1.5 < r < 10 Å. The difference curve (green) is shown offset below. (d) Resulting x-ray PDF refinement goodness-of-fit parameter R_w values versus the temperature on x-ray data using tetragonal (blue) and cubic (red) MgTi₂O₄ models from 90 to 500 K over the range of 1.5 < r < 10 Å. The vertical gray dashed line at 250 K indicates the MIT temperature.

tetragonal model to the 100 K data set is shown in Fig. 2(c). This time, the temperature dependence of the goodness-of-fit parameter R_w of the tetragonal and cubic models [Fig. 2(d)] indicates that the tetragonal model fits the local structure better than the cubic model at all temperatures. In the local structure the tetragonal distortion is always evident.

To see visually in the PDF how the cubic model fails, we consider the low- and high-temperature PDFs and fits with the cubic and tetragonal models, shown in Fig. 3. The lowtemperature PDFs (below the MIT) are shown in Figs. 3(a) and 3(b) and the high-temperature PDFs (above the MIT) in Figs. 3(c) and 3(d). The tetragonal model fit is shown in Figs. 3(a) and 3(c) and the cubic fit is shown in Figs. 3(b) and 3(d). At low temperature, as expected, the tetragonal model [Fig. 3(a)] fits much better than the cubic model [Fig. 3(b)]. At 300 K we have already established that the global structure is cubic; however, again we see that in these local structural fits the tetragonal model [Fig. 3(c)] is superior to that of the cubic model [Fig. 3(d)], with smaller oscillations in the difference curve. To emphasize our argument, we note that the difference curves are similar when data collected at distinct temperatures are fit using identical symmetry constraints. Specifically, careful inspection of the difference curves in Figs. 3(b) and 3(d) reveals that the positions of residual maxima and minima



FIG. 3. The x-ray PDF data (blue) collected at the PDF beamline at (a) and (b) 90 K and (c) and (d) 300 K fit by (a) and (c) tetragonal and (b) and (d) cubic models (red) over the range of 1.5 < r < 4 Å. The difference curves (green) are shown offset below.

are nearly identical, although their amplitudes are smaller in Fig. 3(d). Since this difference signal represents the inability of the cubic model to fit the data, it is further support to the idea that a similar tetragonally distorted structure is present in the low-*r* region at 90 K and at 300 K, but smaller in amplitude at 300 K. To validate that the results are reproducible, Fig. 3 represents x-ray PDF data collected at the PDF beamline, and data collected from the XPD beamline is shown in Fig. 8 presented in the Appendix, which reproduces the same result as discussed above.

In the cubic MgTi₂O₄ structure unit cell, all six Ti-Ti bonds have the same length, 3.008 Å, whereas in the tetragonal MgTi₂O₄ model, one of the six Ti-Ti distances is shortened (2.853 Å), forming dimers, and one Ti-Ti bond is longer (3.157 Å) [44]. The dimer contact is considerably shorter than the 2.917 Å found in titanium metal [45], indicating a strong covalent interaction in MgTi₂O₄. From the analysis of the average structure, it was understood that the MIT was accompanied by the formation of Ti-Ti structural dimers. Above we showed the observation from the PDF of local tetragonality at high temperature, which may indicate that local dimers survive above the MIT. This would seem to be in qualitative agreement with a prior observation from a neutron PDF study that reported the persistence of Ti-Ti dimers up to high temperature [25]. In this picture the dimers survive locally but become disordered at the transition where the structure becomes globally cubic on warming. However, our PDF analysis clearly shows that the large-amplitude structural dimers actually do disappear at the MIT, as described below, and that the local tetragonality that we observe at high temperature in the PDF has a more subtle origin as we develop in greater detail below.



FIG. 4. Simulated (a) x-ray and (b) neutron PDFs of tetragonal (blue) and cubic (red) $MgTi_2O_4$ models in the low-*r* region. All atoms use the same isotropic atomic displacement parameter $U_{iso} = 0.005 \text{ Å}^2$. Here Q_{max} , Q_{damp} , and Q_{broad} are set as the same values as the experimental PDFs. The difference curves (tetragonal model simulation minus cubic model simulation) are shown offset below in green. The nearest Ti-Ti bonds are highlighted in the blue vertical span region. Also shown are the experimental (c) x-ray and (d) neutron PDFs from $MgTi_2O_4$ collected at 100 K (blue) and 300 K (red). The difference curves (100 K data minus 300 K data) are shown offset below in green. The vertical dashed lines at r = 3.01 Årepresent the length of the undistorted Ti-Ti bond length in the average cubic model.

2

3

4

r (Å)

5

2

3

r (Å)

C. Orbital degeneracy lifting or spin dimerization?

To establish the disappearance of the structural dimers at the phase transition, we simulated x-ray PDFs of the cubic (no dimers) and tetragonal (dimers) models. These are plotted as the red and blue curves, respectively, in Fig. 4(a). A number of the PDF peaks are affected on the transition through the MIT, with the largest change observed on the peak at around 3.0 Å, which contains the shortest Ti-Ti distances. Based on the change in crystal structure, the expected change in the PDF results in the characteristic M-shaped signature in the difference curve seen in Fig. 4(a) which comes from the disappearance of the long and short Ti-Ti bonds associated with the dimer, leaving all the Ti-Ti distances the same. This can be compared to the difference in the measured PDF data as MgTi₂O₄ crosses the MIT, shown in Fig. 4(c) at 300 K (red) and 100 K (blue). First, we see a significant change in the relevant 3.0-Å peak when moving through the MIT. This change is inconsistent with the conclusion from the previous neutron PDF study that the dimers are retained, unaltered, to high temperature in the local structure [25], necessitating an order-disorder scenario in which the local structure across the transition is unchanged, with small or no change in the PDF. Instead we see a significant change in the x-ray PDF at the phase transition, with PDF intensity moving from the short r position towards that of the average Ti-Ti distance, as indicated by the gray circle in Fig. 4(c).

It is clear from careful inspection that the observed and simulated difference curves across the transition [Figs. 4(c) and 4(a), respectively] are significantly different. The PDF peak intensity redistribution associated with the dimer removal is seen in the differential curve as a transfer of intensity from 2.71 to 2.89 Å, i.e., on the low-r side of the average Ti-Ti distance at 3.01 Å marked by the vertical dashed line in the figure. This implies that the very short Ti-Ti ion dimers disappear at the phase transition, but the associated PDF intensity is shifted to a position that is still shorter than that of the average Ti-Ti distance: Shortened Ti-Ti contacts in fact do exist at high temperature above the MIT but they are not the original spin-singlet dimers. Our results are consistent with the disappearance of dimers at the phase transition on warming, but a persistence of a local tetragonality, smaller in magnitude than that associated with the the dimer phase. This behavior is shown on an expanded scale in Fig. 5(b), where PDFs measured at multiple temperatures are compared, focusing on the peak at around 3.0 Å. The feature in the difference curve that shows the shift of the intensity associated with the Ti-Ti short bonds from 2.71 Å to the longer position (2.89 Å) is shown below, shaded green (for the loss of short bonds) and purple (for the gain in longer bonds). This result is qualitatively supported by the PDF structural modeling over the range of 1.5 < r < 10 Å, shown in Figs. 5(e) and 5(f), where the short Ti-Ti bond shifts from 2.83 to 2.88 Å over the same temperature range, but never converges to the average cubic value of 3.01 Å at 300 K.

The full temperature dependence of the dimer disappearance may then be extracted by integrating the shaded regions in the difference curve and is shown in Fig. 5(c), where we plot the integrated area shown shaded in green in the difference curve vs temperature. In this case the difference is always taken with respect to the 500 K data set. The intensity decreases gradually until around 200 K, where it rapidly falls off. The rate of falloff in this intensity then slows again above 250 K. There are two principal contributions to the differential signal below the transition: the changes in the local structure and the thermal broadening effects. Thermal broadening effects in the differential are gradual and typically small, particularly from one data point to the next, corresponding to the trends observed below approximately 200 K and above approximately 250 K. On the other hand, the signal in the differential coming from the dimers is considerably larger, roughly proportional to the big step seen at the transition, and dissipates rapidly as the temperature passes through the MIT range. This is a model-independent way of observing how the local dimer disappears on warming.

D. X-ray PDF versus neutron PDF

We may seek an explanation for why we see the dimers disappear at the transition from our x-ray PDF measurements, whereas this was not evident in the earlier neutron PDF study [25]. Observing the local Ti dimer is complicated in the neutron case by the relatively weak and negative neutron scattering length of titanium and an appreciable overlap of titanium contributions with strong oxygen contributions. To



FIG. 5. (a) PDFs measured as a function of temperature from 90 K (blue) to 300 K (red) in the low-r region. The intermediate data sets are plotted in gray. The vertical green arrow at r = 2.71 Å and the purple arrow at 2.89 Å represent how the position of the Ti-Ti short bond changes on warming. The difference curve shown offset below is that between the 90 K data set subtracting the 300 K data set. The positive/negative feature between 2.6 < r < 3.0 Å indicates that intensity in the PDF at 2.71 Å at low temperature is shifting to the position 2.89 Å at high temperature. (b) Re-plot of (a) over an expanded r range. The vertical dashed line at r = 3.01 Å represents the length of the undistorted Ti-Ti bond length in the cubic average structure. (c) Temperature dependence of the integrated area in the difference curve shown shaded green. The vertical gray dashed line at 250 K indicates the MIT temperature. (d) The x-ray PDF refinement goodness-of-fit R_w values versus a variable r_{\min} (for an r_{max} set equal to 50 Å) when fitting a cubic MgTi₂O₄ model to 300 K (blue), 400 K (green), and 500 K (red) data. In each refinement r_{min} was allowed to vary from $1 \leq r_{\min} \leq 36$ Å in 0.2-Å r steps. The fits with high r_{\min} yield the behavior of the average structure. The fits with low r_{\min} are weighted by the local structural signal. Also shown are (e) the temperature dependence of the short (blue) and long (red) Ti-Ti bond lengths and (f) their differences (green) from PDF structural modeling using the tetragonal structure over the range of 1.5 < r < 10 Å. The horizontal dashed line represents the length of the undistorted Ti-Ti bond length in the cubic average structure. The vertical gray dashed line at 250 K indicates the MIT temperature.

show this, we perform the same comparison that we just made for the x-ray PDFs, but on neutron data. The neutron simulations are shown in Fig. 4(b) and the neutron experimental PDFs in Fig. 4(d). It is clear that the signals in the difference curve, both for the simulations based on the average structure and the data PDFs themselves, are much smaller than for the x-ray case. This is because the most important signal in MgTi₂O₄ is coming from the Ti ions that are relatively strong scatterers in the x-ray case but not in the neutron measurement. Specifically, in the x-ray case, Ti scatters 2.75 times stronger than O and 1.8 times stronger than Mg, whereas for neutrons the scattering of Ti is 1.7 times weaker than that of O and 1.6 times weaker than that of Mg. In addition, the neutron PDF intensity in the range of interest is dominated by the contributions from O-O pairs constituting TiO₆ octahedra, evidenced in the data as the additional shoulder intensity features around the 3-Å peak [Fig. 4(d)], as compared to the x-ray case where such features are largely absent [Fig. 4(c)]. This may explain the different interpretation in the earlier neutron PDF study [25]. However, the x-ray data unambiguously show the disappearance of the full-amplitude spin-singlet Ti-Ti dimers.

E. Detection of the ODL state

This behavior, where sizable distortions associated with ordered spin-singlet dimers evolve into smaller local distortions with spin singlets disassembled, is reminiscent of that observed in the CuIr₂S₄ system. In that case a local symmetry-broken ODL state was observed up to the highest temperatures studied [16]. On cooling these local distortions of broken symmetry ordered into a long-range orbitally ordered state. Only below the ordering transition did charges disproportionate, forming a charge density wave accompanied by a Peierls distortion and the formation of Ir-Ir dimers with very short Ir-Ir bonds. We believe our observations in MgTi₂O₄ suggest a similar kind of ODL behavior, which we explore to a greater extent below, though different in detail because of the different charge filling.

As discussed above, we can rule out that the local structure is changing in the same way as the average structure. The *expected* changes in the PDF at low *r* due to average structure changes at the MIT make all the low-*r* peaks sharper for $T > T_{\text{MIT}}$, as shown in Fig. 4(a). However, the data do not show this [Fig. 4(c)]. There is a small change in the local structure, evidenced by the feature in the difference curve around 2.9 Å, indicated by the gray circle in Fig. 4(c), but it is smaller than the average structure change at the MIT. As discussed above, the short Ti-Ti dimers (r = 2.71 Å shoulder) go away on warming by a shift to a longer bond (around r = 2.89 Å), but this "longer" bond is still shorter than the average 3.01-Å Ti-Ti bond distance expected from the cubic average model. These two behaviors exactly mimic the ODL state found in CuIr₂S₄ [16].

As is evident in Fig. 2(d), the tetragonal model fits the local structure better than the cubic model at all temperatures to 500 K, the highest measurement temperature. As in $CuIr_2S_4$ [16] the dimers disappear at the MIT transition on warming, but the local symmetry-broken state with, presumably, fluctuating short Ti-Ti bonds is present to high temperature.

F. Spatial extent of the ODL state

It is of interest to explore whether the fluctuating short Ti-Ti bonds at temperature above the MIT correlate with each other and how this varies with temperature. To extract the correlation length of the local fluctuating symmetry-broken states, $r_{\rm min}$ -dependent PDF fits (see Sec. II) were performed on selected high-temperature data sets (300, 400, and 500 K).

When r_{\min} is large the fit is over just the high-*r* region of the PDF and will retrieve the average structure and the cubic fit will be good. As r_{\min} extends to lower values, progressively more of the local structure that has a tetragonal distortion is included in the fit and the agreement of the cubic model becomes degraded. The resulting $R_w(r_{\min})$ is shown in Fig. 5(d). The cubic fits are good over the entire region $r_{\min} > 10$ Å, with very little variation in R_w indicated by the horizontal gray band in the figure, but rapidly degrade below this length scale. This suggests that the symmetry-broken local distortions have a correlation length of around 1 nm but that this correlation length does not vary significantly in the temperature range above 300 K. This agrees well with the previously reported length scale of local tetragonality based on neutron PDF analysis [25].

IV. TWO-ORBITAL ODL STATE

A. The ODL regime

The driving force behind the high-temperature local symmetry breaking in $CuIr_2S_4$ was shown to be of electronic origin involving orbital degeneracy lifting [16]. Significantly, the isostructural and isoelectronic sister compound $CuIr_2Se_4$ remains metallic down to the lowest temperature and does not show local symmetry-breaking ODL effects. This intriguingly suggests that the ODL state could be *a prerequisite* for the MIT [16] in these spinels. Here we argue that also in MgTi₂O₄ local ODL effects are present and produce the local symmetry breaking in the high-temperature region that we report here. By analogy with CuIr₂S₄, the ODL states are precursor states to the spin-singlet dimerization and MIT.

The long-range pattern of ordered spin-singlet dimers differs in detail in the two systems, being octamers in $CuIr_2S_4$ [2] and helices in MgTi₂O₄ [3], and is presumably dictated by delicate energetics [46,47]. Although intriguing, these intricate aspects of long-range order are not of concern here. Our interest lies with internal dimer structure, the relationship between the dimer state and the ODL state, and the electronhole symmetry that links the two systems [21].

Prior to pursuing the analogy between CuIr₂S₄ and MgTi₂O₄, we appraise the two systems from the perspective of a regime under which the ODL state can form in a transition-metal system [16]. Three criteria need to be fulfilled: (i) filling, where the transition metal in the system possesses partially filled d orbitals; (ii) symmetry, where the high crystallographic point symmetry of the system imposes the orbital degeneracy; and (iii) topology, where the structural topology promotes adequate highly directional orbital overlaps. In practice, materialization of the ODL state could further be impacted by the existence of other competing degeneracy lifting channels that may be available to the system, such as the relativistic spin orbit coupling, the effects of crystal field, etc. Eventually, formation of the ODL local broken-symmetry state is accompanied by associated local structure distortion. Both CuIr₂S₄ and MgTi₂O₄ systems meet the criteria: partial filling (Ir is $5d^{5.5}$ and Ti is $3d^{1}$), degeneracy-promoting high-symmetry structure (cubic spinel structure imposes threefold degeneracy of the t_{2g} manifolds in both materials), and favorable structural topology (edge



FIG. 6. Recapitulation of the ODL mechanism in the CuIr₂S₄ thiospinel. (a) Energy diagram of atomic and molecular orbitals for nearest Ir atom pairs in various situations: degenerate/nonbonding (DEG/NONBOND, top), orbital degeneracy lifted (ODL, middle), and spin-singlet dimer (DIMER, bottom). Horizontal arrows indicate the ODL displacements. (b) The ODL state in $CuIr_2S_4$ in the t_{2g} orbital manifold representation on a segment of pyrochlore sublattice. The ODL active t_{2g} orbitals are shown in yellow and passive orbitals are shown in gray. Each Ir atom participates in exactly one ODL state. The ODL states are randomly placed following principles discussed in the text. The arrows denote the displacements of Ir associated with the ODL states and provide mapping onto the two-in-two-out ice rules. The white arrows represent in-plane displacements, whereas the blue arrows represent displacements with an out-of-plane component. The dotted arrows mark the displacements associated with the ODL states occurring in neighboring tetrahedra that are not represented in the displayed structural segment.

shared IrS₆ and TiO₆ octahedra fostering direct t_{2g} overlap). In both systems short nearest-neighbor transition-metal contacts are observed in the high-temperature metallic regime that are distinct from the very short spin-singlet dimer bonds observed in the insulating state at low temperature.

In what follows we first review the ODL state and the dimerization mechanism within the local picture of $CuIr_2S_4$, summarized in Fig. 6. We then utilize the analogies between the two systems to put forward a scenario portraying the high-temperature state and the dimerization mechanism in $MgTi_2O_4$, illustrated in Fig. 7.

B. Single-orbital ODL state

In CuIr₂S₄ the dimers involve two Ir⁴⁺ ions in 5d⁵ configuration with hole character [16]. In the localized electron picture, the ODL state is based on molecular orbital (MO) concepts. In this, two neighboring transition-metal ions with partially filled degenerate orbitals form a MO state with shared electrons (holes) that lifts the degeneracy and lowers the system energy. In the high-temperature regime of CuIr₂S₄ iridium nominally has a half-integer valence (3.5+) [11,48]. This means that each Ir^{3.5+} on the pyrochlore sublattice is in a nominal 5d^{5.5} state corresponding to half a hole per three degenerate t_{2g} orbitals [Fig. 6(a), top]. The sublattice geometry fosters direct overlaps of orbitals from the t_{2g} manifolds [Fig. 6(b)], which in turn allows neighboring Ir pairs to form a bound state sharing a single hole in the antibonding MO [16] [middle panel of Fig. 6(a)].

Importantly, for any choice of two neighboring Ir on a pyrochlore lattice only one member of the three t_{2g} orbitals



FIG. 7. The Ti t_{2g} orbital manifold on a pyrochlore sublattice of the MgTi₂O₄ spinel: (a) spin-singlet dimerized lattice within the tetragonal model, and (c) lattice of degenerate t_{2g} orbitals as portrayed by the cubic model. The charge state of all Ti in all models is +3. The electron density in t_{2g} orbitals is indicated by color (gray is empty) and its distribution is proportional to the blue color intensity. The Ti dimers in (a) are indicated by the double blue lines, while a pair of antiferromagnetically coupled block arrows on an exemplar dimer denotes its spin-singlet character. Note that all Ti atoms are involved in dimerization but only dimers contained in a section of one structural slab are indicated in (a). Out of all Ti-Ti contacts, dimerized contacts have a nominal bond charge of $2e^-$, whereas the other contacts carry no net charge in this picture. The ODL states in (b) are indicated by the model discussed in the text. The insets between the panels provide the energy diagram of atomic and molecular orbitals for nearest Ti atom pairs in various situations: degenerate/nonbonding (DEG/NONBOND, bottom right), spin-singlet dimer (DIMER, top left), one-orbital ODL (10-ODL, top right), and two-orbital ODL (20-ODL, bottom left). Note that the 20-ODL state refers to two orbitals of the same Ti (e.g., one labeled 2) that are engaged in two 10-ODL states with two *different* Ti neighbors (e.g., those labeled 1 and 3). Differently colored spins and lines denoting ODL contacts indicate this two-component aspect in the schematics for 20-ODL. The local spin arrangement shown is for illustration purposes only and does not represent an experimentally established alignment.

overlap (e.g., xy with xy, etc.) along the Ir₄ tetrahedral edges. Due to the specifics of filling (0.5 hole/Ir) each Ir participates in exactly one such paired state at a time. We call this a one-orbital ODL (10-ODL) state. The ODL state is hence comprised of two atomic orbitals, one from each Ir in the pair, with on average 1.5 electrons (0.5 hole) per Ir, resulting in MO with three electrons and one hole, as shown in Fig. 6(a), with a net spin of 1/2. This configuration results in the observed contraction of the Ir-Ir separation in the local structure vis-àvis that expected if orbital degeneracy is retained. Since each iridium has six Ir neighbors to pair with, the ODL state fluctuates spatially and, presumably, temporally among (xy, xy), (yz, yz), and (zx, zx) variants, which results in an undistorted cubic structure on average [16]. One such configuration is illustrated in Fig. 6(b), with strong short-range correlations governed by the one ODL state per Ir and the Coulomb bond charge repulsion principles, resulting in a single ODL state per Ir₄ tetrahedron. Pursuant to this, the ODL state in CuIr₂S₄ follows the two-in-two-out ice rules [49], which will be addressed later.

In this localized picture, the ODL state is also a precursor for the spin-singlet dimer. The dimer state is attained by removing an excess electron from the antibonding MO of the ODL state, thus stabilizing the bond, as shown in the bottom panel of Fig. 6(a). The process involves charge transfer between two ODL $Ir^{3.5+}-Ir^{3.5+}$ pairs, one of which becomes dimerized $Ir^{4+}-Ir^{4+}$ by losing an electron (or gaining a hole, hence hole dimer) and the other becomes non-ODL (and nondimer) $Ir^{3+}-Ir^{3+}$ by gaining an electron (d^6-d^6 configuration).

C. Two-orbital ODL state

We now turn to MgTi₂O₄ starting from the local view of spin-singlet dimers. Dimerization in MgTi₂O₄, depicted using the Ti t_{2g} manifold representation, is shown in Fig. 7(a), overlaying a fragment of pyrochlore sublattice as seen in the $P4_12_12$ model. In MgTi₂O₄ the dimers involve two Ti³⁺ in the $3d^1$ configuration with a single electron in the t_{2g} manifold; hence Ti³⁺-Ti³⁺ dimers inevitably have electron character by construction. These result in short Ti-Ti dimerized contacts observed in the tetragonal structure at low temperature. Each dimer carries $2e^{-}$ of net charge. Since each Ti participates in a dimer and since there is exactly one dimer per Ti₄ tetrahedron [25,46], nominal charge count results in $1e^{-1}$ Ti site. This is consistent with no CO being observed experimentally in MgTi₂O₄, in contrast to CuIr₂S₄, implying that all Ti sites are equivalent in this regard [3]. From the average structure perspective and in the localized picture, supported by the experimentally observed paramagnetism and poor metallic conduction, above the MIT the dimers could be seen to disassemble in such a way as to statistically distribute $1e^-$ of charge evenly across the three degenerate t_{2g} orbitals, resulting in a cubic structure with all Ti-Ti nearest-neighbor contacts equivalent, as schematically presented in Fig. 7(c). This implies that, despite nominal charge equivalence of all Ti sites and no site CO observed, some charge transfer, presumably involving bond charge, still has to take place at the transition. This then inevitably implies that the ground state has to involve the bond charge order which coincides with, and is hence indistinguishable from, the observed dimer order in the diffraction measurements.

The implication of charge being equally distributed across the triply degenerate t_{2g} manifold of Ti in a manner depicted in Fig. 7(c), presented also in an associated energy diagram (bottom right inset of the figure), would be that the pyrochlore sublattice is comprised of regular Ti₄ tetrahedra with equidistant Ti-Ti contacts corresponding to equal bond charge, as described within the cubic spinel model. However, this is not what is observed experimentally. The observations based on the PDF analyses clearly demonstrate that the pyrochlore sublattice is comprised of locally distorted Ti₄ tetrahedra with a distribution of distances, suggesting that the bond charge remains inequivalent above the MIT. One possibility is that the dimers indeed persist in the metallic regime, as hinted at in the earlier neutron study [25]. However, such an interpretation would be inconsistent with magnetization measurements of MgTi₂O₄ that establish the disappearance of spin singlets above the MIT. It would further be inconsistent with the elongation of the short Ti-Ti dimer contacts evidenced in our x-ray PDF analysis, which implies the disappearance of spin-singlet dimers at MIT even locally. In analogy with CuIr₂S₄, it is plausible that the dimer state gets replaced locally by an ODLtype state in the high-temperature metallic phase. Observation of heterogeneous local Ti-Ti contacts corresponding to inequivalent bond charge is consistent with an ODL-like state in MgTi₂O₄. Since the local structures at high temperature and in the ground state are distinct, the MIT in MgTi₂O₄ cannot be assigned to a trivial order-disorder type.

Importantly, as careful assessment shows, the ODL state in MgTi₂O₄ cannot be exactly mapped onto the ODL state seen in CuIr₂S₄, since the charge filling is different in the two systems $(1e^{-}/\text{Ti} \text{ in MgTi}_2\text{O}_4 \text{ and } 0.5 \text{ hole}/\text{Ir in CuIr}_2\text{S}_4)$. In the latter case two Ir neighbors can reduce the energy of the system by forming a 10-ODL state accommodating a common hole, as described in the preceding section. This results in a three-electron state shown in the middle panel of Fig. 6(a). Given that the destruction of the spin-singlet dimer in MgTi₂O₄ involves removal of one of the electrons from the dimer state sketched in the energy diagram in the top left inset of Fig. 7, consistent with destabilization and elongation of the short Ti-Ti bond, in the ODL picture the one electron left behind would indeed result in an electron-hole antipode of the 10-ODL state seen in CuIr₂S₄, as shown in the top right inset of Fig. 7. The issue arises with the placement of the extra electron from the dimer. In the case of CuIr₂S₄, due to filling, the removal of one of the two holes constituting a dimer results in generation of two 10-ODL states on two different pairs of Ir. There, since only 50% of Ir participates in dimerization, in terms of filling one should think of this process as a replacement of the 4+-4+-3+-3+ charge tetramer with a pair of 3.5+-3.5+ ODL states achieved by hole redistribution. Since there is no site charge disproportionation in MgTi₂O₄ and since each Ti is involved in dimerization, the dimer density per formula unit is twice as large in MgTi₂O₄ as in CuIr₂S₄ and each Ti has to participate in two independent 10-ODL states simultaneously. Notably, the geometry of t_{2g} orbital overlaps imposes a constraint that the two 1O-ODL states for each Ti have to be assembled with two *different* Ti neighbors, as illustrated in Figs. 7(a) and 7(b). There, for example, the dimer in Fig. 7(a) involving Ti labeled 2 in Fig. 7(b) disassembles at the MIT to make two 10-ODL states, one with the Ti neighbor labeled 1 and another with the neighbor labeled 3. This results in two short Ti-Ti contacts that are both longer than the dimer distance but shorter than the average Ti-Ti separation in the cubic structure. We call this a two-orbital ODL (20-ODL) state, given that two atomic orbitals of a single Ti ion are utilized. Such a 20-ODL state is hence comprised of a superposition of two 10-ODL states of different variety [e.g., (xy, xy) and (yz, yz), (yz, yz) and (zx, zx), etc.] that point in different directions and lie along different edges of the pyrochlore sublattice. These ODL distances are marked as thick lines color coded as red and blue in Fig. 7(b), and the corresponding state is schematically shown on the energy diagram in the bottom left inset of the figure, where the matching color coding of the electron spins signifies that they belong to different 10-ODL component states.

The existence of the 2O-ODL state in MgTi₂O₄ is corroborated by another experimentally observed difference between the two systems: The spatial extent of the local order associated with the metallic regime of MgTi₂O₄ is observably larger than that seen in $CuIr_2S_4$ [16]. The extended character of the local structural distortions associated with the 2O-ODL state in MgTi₂O₄ is expected for the following reasons. First, due to filling, the two ingredient 10-ODL prong states in the 20-ODL superstate cannot both be of the same type [e.g., if one is (xy, xy), the other can only be (yz, yz) or (zx, zx)], which would tend to increase local structural correlations. Second, the bond charge on different 10-ODL bonds is of the same sign, resulting in the Coulomb repulsion, which would also maximize the span of the 2O-ODL state itself. The observed local distortions over the length scale of about 1 nm, spanning approximately three Ti₄ tetrahedra [25], are therefore consistent with the presence of the 2O-ODL state in MgTi₂O₄.

The energetic benefit of the 2O-ODL state over the local spin-singlet dimer state is not apparent and remains elusive. The local 2O-ODL and disordered dimer states would both increase the system entropy and in turn the entropic contribution to energy would stabilize corresponding shortrange-order state at elevated temperature. On the other hand, some vestigial magnetic correlations could be expected in the high-temperature ODL regime. Another possibility then is that the principal stabilization of the 2O-ODL state stems from its presumed magnetism. The character of local spin correlations within the 2O-ODL state cannot be established by the analysis carried out in this work. In the 2O-ODL diagram shown in Fig. 7 the two spins are arbitrarily drawn as parallel to avoid any confusion with the dimer state, but their relationship has actually not been established experimentally. In fact, the magnetic response of $MgTi_2O_4$ above the MIT [Fig. 1(c)] is neither Pauli-like nor Curie-Weiss-like. Rather it resembles that of charge density wave systems at high temperature [50], the regime associated with a pseudogap in the electronic density of states [51]. In these systems magnetic susceptibility behavior at $T > T_s$ was attributed to fluctuations of the charge density wave amplitude [52,53]. It is thus tempting to speculate that in MgTi₂O₄ the observed magnetic response may similarly be due to ODL fluctuations. It would therefore be of appreciable interest to explore this

aspect of the 2O-ODL state by techniques sensitive to local magnetism, such as muon spin rotation [54] and magnetic PDF [55], which would be particularly informative in that regard.

While the PDF probe used here provides the information on instantaneous atomic structure and as such does not differentiate between static and dynamic disorder, the ODL state in these systems is expected to be dynamic. Spatiotemporal fluctuations then average out to the perceived undistorted cubic average structure as observed crystallographically. Notably, the resistivity just above the MIT in $CuIr_2S_4$ is about 2 m Ω cm and linearly increasing with temperature [14], ascribed to a bipolaronic hopping mechanism [15,48], whereas in the metallic regime of MgTi₂O₄ not only is electric resistivity substantially higher, but it decreases with increasing temperature in an insulatorlike manner [8]. This stark difference in the observed electronic transport could be considered as an important indicator, albeit indirect, of the underlying difference reflecting the 10-ODL and 20-ODL characters of the high-temperature states in these two systems, respectively.

Although the Ir dimers in CuIr₂S₄ are strictly speaking equivalent to the Ti dimers in MgTi₂O₄, the mechanism of their local formation from the ODL state is electron-hole symmetric and in that sense the dimers in these two systems could be considered as having a different flavor derived from their origin. Formation of dimers in CuIr₂S₄ requires transfer of holes from one half of the available population of ODL states to the other, and the ODLs receptors of a hole become dimers, accounting for only 50% of Ir being dimerized. In contrast, the dimers in MgTi₂O₄ assemble from the ODL states by a virtue of electron transfer, where 10-ODL states that receive electrons become dimers with all Ti participating in dimerization. While in both systems the process involves bond charge disproportionation, in CuIr₂S₄ this consequentially results in the observed site charge disproportionation and subsequent charge order, which is presumably imposed by the specifics of the filling and reflected in the dimer density per formula unit.

D. Consequences for ice-type nanoscale fluctuations

The presence of the 2O-ODL state has another important consequence for MgTi₂O₄. Each Ti₄ tetrahedron inevitably hosts two 10-ODL states. Due to the Coulomb repulsion of the bond charge, we would expect the two states to be placed on the opposite skew edges of each tetrahedron, although other constellations cannot be excluded. On the other hand, irrespective of the details of their distribution, multiple 10-ODLs on one Ti₄ tetrahedron would cause distortions that are incompatible with the ice-type structural fluctuations in the MgTi₂O₄ system such as those suggested in the previous study of the local structure of MgTi₂O₄ [25]. Given that the spin-singlet dimer distortion in the ground state of MgTi₂O₄ [3] follows the two-in-two-out ice rules [49] on each individual Ti₄ tetrahedron in the structure, the proposition that the local Ti atomic displacements have the same configuration in the cubic phase [25] presumably originates in part from the order-disorder-type view of the MIT in MgTi₂O₄ that would be implicated by the survival of dimerlike distortions in the high-temperature phase. Our analysis does not support this picture. However, based on the considerations

described above, CuIr_2S_4 may possibly be a better candidate for exhibiting the distortions of the two-in–two-out type in the disordered ODL regime, as illustrated by the mapping shown in Fig. 6(b). Exploring this matter further both experimentally and theoretically should provide a more detailed understanding of these systems. Single-crystal diffuse-scattering-based methods [56–60] and dynamical mean-field and advanced first-principles approaches [61–63] would be particularly useful in that regard.

V. CONCLUSION

Here we applied joint x-ray and neutron pair distribution function analysis on the dimerized MgTi₂O₄ spinel, a candidate system for hosting the multiorbital ODL state, to track the evolution of its local atomic structure across its localized-toitinerant electronic transition. Consistent with recent reports, the local structure does not agree with the average structure above the MIT temperature of 250 K and deep in the metallic cubic regime. However, in stark contrast to previous findings [25], we provide unambiguous evidence that spinsinglet dimers are vanishing at the MIT. The shortest Ti-Ti distance corresponding to spin-singlet dimers experiences a discontinuous elongation locally on warming through the MIT but remains shorter than that prescribed by the cubic average structure. The local distortion in the metallic regime is quantitatively and qualitatively different from that observed in association with the spin-singlet state, implying that the MIT is not a trivial order-disorder-type transition. The distortion characterizes the entire metallic regime and persists up to at least 500 K (about $2T_s$). The observed behavior is an identifying characteristic of the local symmetry-broken ODL state observed in the related CuIr₂S₄ system. The correlation length of local distortions associated with the ODL state in MgTi₂O₄ is about 1 nm, which is double that seen in $CuIr_2S_4$, implying two-orbital character of the ODL state. The observations exemplify that high-temperature electronic precursor states that govern emergent complex low-temperature behaviors in quantum materials can indeed have a multiorbital ODL character.

The Department of Energy will provide public access to these results of federally sponsored research in accordance with the DOE Public Access Plan [64].

ACKNOWLEDGMENTS

Work at Brookhaven National Laboratory was supported by the U.S. Department of Energy (DOE), Office of Science, Office of Basic Energy Sciences under Contract No. DE-SC0012704. L.Y. and M.G.T. acknowledge support from the ORNL Graduate Opportunity program, which was funded by the Neutron Science Directorate, with support from the Scientific User Facilities Division, Office of Basic Energy Science, U.S. DOE. Work in the Materials Science Division at Argonne National Laboratory (sample synthesis and characterization) was sponsored by the U.S. DOE Office of Science, Basic Energy Sciences, Materials Science and Engineering Division. X-ray PDF measurements were conducted at 28-ID-1 and 28-ID-2 beamlines of the National Synchrotron Light Source II, a U.S. DOE Office of Science User Facility operated for the DOE Office TABLE I. Rietveld refinement results of tetragonal and cubic MgTi₂O₄ models fit to the neutron TOF powder diffraction patterns at 100 and 500 K, respectively. The models are introduced in detail in Sec. II. Here R_{wp} is the weighted profile agreement factor; *x*, *y*, and *z* are the refinable atomic positions in fractional coordinates; U_{iso} , in units of Å², is the isotropic atomic displacement parameter.

Parameter	Model (Temperature)	
	Tetragonal (100 K)	Cubic (500 K)
$\overline{R_{wp}}$	0.06534	0.05154
a (Å)	6.01197	8.50833
<i>c</i> (Å)	8.46956	-
$c/\sqrt{2}a$	0.99616	1.0
Mg x	0.7449	-
Ti(x, y, z)	(-0.0112, 0.2451, -0.1414)	-
O1 (x, y, z)	(0.4795, 0.2480, 0.1182)	(0.2595, 0.2595, 0.2595)
O2 (x, y, z)	(0.2410,0.0219,0.8805)	-
Mg $U_{\rm iso}$ (Å ²)	0.0063	0.0110
Ti $U_{\rm iso}$ (Å ²)	0.0093	0.0191
$O U_{iso} (Å^2)$	0.0042	0.0071

of Science by Brookhaven National Laboratory. Neutron diffraction experiments were carried out at the NOMAD beamline of the Spallation Neutron Source, Oak Ridge National Laboratory, which was sponsored by the Scientific User Facilities Division, Office of Basic Energy Science, U.S. DOE.

APPENDIX: SUPPLEMENTAL RESULTS

Structural parameters obtained from Rietveld refinements of the tetragonal and cubic models fit to 100 and 500 K



FIG. 8. The x-ray PDF data (blue) collected at the XPD beamline at (a) and (b) 90 K and (c) and (d) 300 K fit by (a) and (c) tetragonal and (b) and (d) cubic models (red) over the range of 1.5 < r < 4 Å. The difference curves (green) are shown offset below.

PHYSICAL REVIEW B 102, 235128 (2020)

neutron time-of-flight powder diffraction data, respectively, are summarized in Table I. These confirm the expected average structure behavior of our sample and are in good quantitative agreement with the values reported previously [3].

The data collected in a separate x-ray PDF experiment carried out at a different beamline under experimental conditions similar to those of the main x-ray PDF experiment are shown in Fig. 8. The PDF analysis of these data reproduces the observations from the main experiment presented in Fig. 3.

- K. E. Sickafus, J. M. Wills, and N. W. Grimes, J. Am. Ceram. Soc. 82, 3279 (1999).
- [2] P. G. Radaelli, Y. Horibe, M. J. Gutmann, H. Ishibashi, C. H. Chen, R. M. Ibberson, Y. Koyama, Y.-S. Hor, V. Kiryukhin, and S.-W. Cheong, Nature (London) 416, 155 (2002).
- [3] M. Schmidt, W. Ratcliff, P. G. Radaelli, K. Refson, N. M. Harrison, and S. W. Cheong, Phys. Rev. Lett. 92, 056402 (2004).
- [4] M. Matsuda, H. Ueda, A. Kikkawa, Y. Tanaka, K. Katsumata, Y. Narumi, T. Inami, Y. Ueda, and S.-H. Lee, Nat. Phys. 3, 397 (2007).
- [5] P. G. Radaelli, New J. Phys. 7, 53 (2005).
- [6] J. P. Wright, J. P. Attfield, and P. G. Radaelli, Phys. Rev. Lett. 87, 266401 (2001).
- [7] S. Nagata, N. Matsumoto, Y. Kato, T. Furubayashi, T. Matsumoto, J. P. Sanchez, and P. Vulliet, Phys. Rev. B 58, 6844 (1998).
- [8] M. Isobe and Y. Ueda, J. Phys. Soc. Jpn. 71, 1848 (2002).
- [9] M. Ito, J. Hori, H. Kurisaki, H. Okada, A. J. Perez Kuroki, N. Ogita, M. Udagawa, H. Fujii, F. Nakamura, T. Fujita, and T. Suzuki, Phys. Rev. Lett. 91, 077001 (2003).
- [10] T. Furubayashi, T. Matsumoto, T. Hagino, and S. Nagata, J. Phys. Soc. Jpn. 63, 3333 (1994).
- [11] J. Matsuno, T. Mizokawa, A. Fujimori, D. A. Zatsepin, V. R. Galakhov, E. Z. Kurmaev, Y. Kato, and S. Nagata, Phys. Rev. B 55, R15979(R) (1997).
- [12] N. Matsumoto, R. Endoh, S. Nagata, T. Furubayashi, and T. Matsumoto, Phys. Rev. B 60, 5258 (1999).
- [13] H. Ishibashi, T. Sakai, and K. Nakahigashi, J. Magn. Magn. Mater. 226–230, 233 (2001).
- [14] A. T. Burkov, T. Nakama, M. Hedo, K. Shintani, K. Yagasaki, N. Matsumoto, and S. Nagata, Phys. Rev. B 61, 10049 (2000).
- [15] K. Takubo, T. Mizokawa, N. Matsumoto, and S. Nagata, Phys. Rev. B 78, 245117 (2008).
- [16] E. S. Bozin, W. G. Yin, R. J. Koch, M. Abeykoon, Y. S. Hor, H. Zheng, H. C. Lei, C. Petrovic, J. F. Mitchell, and S. J. L. Billinge, Nat. Commun. 10, 3638 (2019).
- [17] E. S. Bozin, A. S. Masadeh, Y. S. Hor, J. F. Mitchell, and S. J. L. Billinge, Phys. Rev. Lett. **106**, 045501 (2011).
- [18] R. J. Koch, T. Konstantinova, M. Abeykoon, A. Wang, C. Petrovic, Y. Zhu, E. S. Bozin, and S. J. L. Billinge, Phys. Rev. B 100, 020501(R) (2019).
- [19] B. A. Frandsen, Q. Wang, S. Wu, J. Zhao, and R. J. Birgeneau, Phys. Rev. B 100, 020504(R) (2019).
- [20] X. Qiu, T. Proffen, J. F. Mitchell, and S. J. L. Billinge, Phys. Rev. Lett. 94, 177203 (2005).
- [21] D. I. Khomskii and T. Mizokawa, Phys. Rev. Lett. 94, 156402 (2005).
- [22] S. Nagata, T. Hagino, Y. Seki, and T. Bitoh, Physica B 194–196, 1077 (1994).

- [23] J. Zhou, G. Li, J. L. Luo, Y. C. Ma, D. Wu, B. P. Zhu, Z. Tang, J. Shi, and N. L. Wang, Phys. Rev. B 74, 245102 (2006).
- [24] M. Croft, W. Caliebe, H. Woo, T. A. Tyson, D. Sills, Y. S. Hor, S.-W. Cheong, V. Kiryukhin, and S.-J. Oh, Phys. Rev. B 67, 201102(R) (2003).
- [25] S. Torigoe, T. Hattori, K. Kodama, T. Honda, H. Sagayama, K. Ikeda, T. Otomo, H. Nitani, H. Abe, H. Murakawa, H. Sakai, and N. Hanasaki, Phys. Rev. B 98, 134443 (2018).
- [26] T. Egami and S. J. L. Billinge, Underneath the Bragg Peaks: Structural Analysis of Complex Materials, 2nd ed. (Elsevier, Amsterdam, 2012).
- [27] S. J. L. Billinge, in *International Tables for Crystallography*, edited by C. J. Gilmore, J. A. Kaduk, and H. Schenk (International Union of Crystallography, Chester, 2019), Vol. H, pp. 649–672.
- [28] S. J. L. Billinge, R. G. DiFrancesco, G. H. Kwei, J. J. Neumeier, and J. D. Thompson, Phys. Rev. Lett. 77, 715 (1996).
- [29] C. A. Young and A. L. Goodwin, J. Mater. Chem. 21, 6464 (2011).
- [30] D. A. Keen and A. L. Goodwin, Nature (London) 521, 303 (2015).
- [31] J. V. Laveda, B. Johnston, G. W. Paterson, P. J. Baker, M. G. Tucker, H. Y. Playford, K. M. Ø. Jensen, S. J. L. Billinge, and S. A. Corr, J. Mater. Chem. A 6, 127 (2018).
- [32] C. L. Farrow and S. J. L. Billinge, Acta Crystallogr. A 65, 232 (2009).
- [33] P. J. Chupas, X. Qiu, J. C. Hanson, P. L. Lee, C. P. Grey, and S. J. L. Billinge, J. Appl. Crystallogr. 36, 1342 (2003).
- [34] T. Proffen and S. J. L. Billinge, J. Appl. Crystallogr. 32, 572 (1999).
- [35] C. L. Farrow, P. Juhás, J. Liu, D. Bryndin, E. S. Božin, J. Bloch, T. Proffen, and S. J. L. Billinge, J. Phys.: Condens. Matter 19, 335219 (2007).
- [36] G. Ashiotis, A. Deschildre, Z. Nawaz, J. P. Wright, D. Karkoulis, F. E. Picca, and J. Kieffer, J. Appl. Crystallogr. 48, 510 (2015).
- [37] P. Juhás, T. Davis, C. L. Farrow, and S. J. L. Billinge, J. Appl. Crystallogr. 46, 560 (2013).
- [38] X. Yang, P. Juhás, C. L. Farrow, and S. J. L. Billinge, arXiv:1402.3163.
- [39] J. Neuefeind, M. Feygenson, J. Carruth, R. Hoffmann, and K. K. Chipley, Nucl. Instrum. Methods Phys. Res. Sect. B 287, 68 (2012).
- [40] P. Juhás, C. L. Farrow, X. Yang, K. R. Knox, and S. J. L. Billinge, Acta Crystallogr. A 71, 562 (2015).
- [41] I. Jeong, T. Proffen, F. Mohiuddin-Jacobs, and S. J. L. Billinge, J. Phys. Chem. A 103, 921 (1999).
- [42] B. H. Toby and R. B. Von Dreele, J. Appl. Crystallogr. 46, 544 (2013).
- [43] A. N. Vasiliev, M. M. Markina, M. Isobe, and Y. Ueda, J. Magn. Magn. Mater. **300**, e375 (2006).

- [44] V. M. Talanov, V. B. Shirokov, V. V. Ivanov, and M. V. Talanov, Crystallogr. Rep. 60, 101 (2015).
- [45] A. W. Hull, J. Franklin Inst. 193, 189 (1922).
- [46] S. Di Matteo, G. Jackeli, and N. B. Perkins, Phys. Rev. B 72, 024431 (2005).
- [47] A. Rahaman, M. Chakraborty, T. Paramanik, R. K. Maurya, S. Mahana, R. Bindu, D. Topwal, P. Mahadevan, and D. Choudhury, Phys. Rev. B 100, 115162 (2019).
- [48] K. Yagasaki, T. Nakama, M. Hedo, Y. Uwatoko, Y. Shimoji, S. Notsu, K. Uchima, N. Matsumoto, S. Nagata, H. Okada, H. Fujii, H. Yoshida, H. M. Kimura, Y. Yamaguchi, and A. T. Burkov, J. Phys. Soc. Jpn. 75, 074706 (2006).
- [49] P. W. Anderson, Phys. Rev. 102, 1008 (1956).
- [50] J. L. Benchimol, F. T. Hedgcock, and F. J. Di Salvo, Solid State Commun. 25, 677 (1978).
- [51] S. V. Borisenko, A. A. Kordyuk, A. N. Yaresko, V. B. Zabolotnyy, D. S. Inosov, R. Schuster, B. Büchner, R. Weber, R. Follath, L. Patthey, and H. Berger, Phys. Rev. Lett. 100, 196402 (2008).
- [52] D. C. Johnston, Phys. Rev. Lett. 52, 2049 (1984).
- [53] J. Zhang, D. Phelan, A. S. Botana, Y.-S. Chen, H. Zheng, M. Krogstad, S. G. Wang, Y. Qiu, J. A. Rodriguez-Rivera,

PHYSICAL REVIEW B 102, 235128 (2020)

R. Osborn, S. Rosenkranz, M. R. Norman, and J. F. Mitchell, Nat. Commun. **11**, 6003 (2020).

- [54] I. McKenzie, Annu. Rep. Prog. Chem. C 109, 65 (2013).
- [55] B. Frandsen, X. Yang, and S. J. L. Billinge, Acta Crystallogr. A 70, 3 (2014).
- [56] T. Weber and A. Simonov, Z. Kristallogr. Cryst. Mater. 227, 238 (2012).
- [57] M. J. Krogstad, S. Rosenkranz, J. M. Wozniak, G. Jennings, J. P. C. Ruff, J. T. Vaughey, and R. Osborn, Nat. Mater. 19, 63 (2020).
- [58] M. A. Davenport, M. J. Krogstad, L. M. Whitt, C. Hu, T. C. Douglas, N. Ni, S. Rosenkranz, R. Osborn, and J. M. Allred, arXiv:1909.12704.
- [59] N. Roth, A. F. May, F. Ye, B. C. Chakoumakos, and B. B. Iversen, IUCrJ 5, 410 (2018).
- [60] N. Roth and B. B. Iversen, Acta Crystallogr. A 75, 465 (2019).
- [61] Y. Pramudya, H. Terletska, S. Pankov, E. Manousakis, and V. Dobrosavljević, Phys. Rev. B 84, 125120 (2011).
- [62] S. Mahmoudian, L. Rademaker, A. Ralko, S. Fratini, and V. Dobrosavljević, Phys. Rev. Lett. 115, 025701 (2015).
- [63] Z. Wang, X.-G. Zhao, R. Koch, S. J. L. Billinge, and A. Zunger, Phys. Rev. B 102, 235121 (2020).
- [64] http://energy.gov/downloads/doe-public-access-plan.